

**M.Sc. Semester-IV
Core Course-9 (CC-9)
Synthetic Organic Chemistry**



**II. Pericyclic Reactions
1. Molecular Orbital Symmetry**



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II Pericyclic Reactions 20 Hrs

Molecular orbital symmetry, Frontier orbitals of ethylene, 1,3-butadiene, 1, 3, 5-hexatriene, allyl system, Classification of pericyclic reactions. FMO approach, Woodward-Hoffman correlation diagram method and PMO approach for pericyclic reaction under thermal and photochemical conditions.

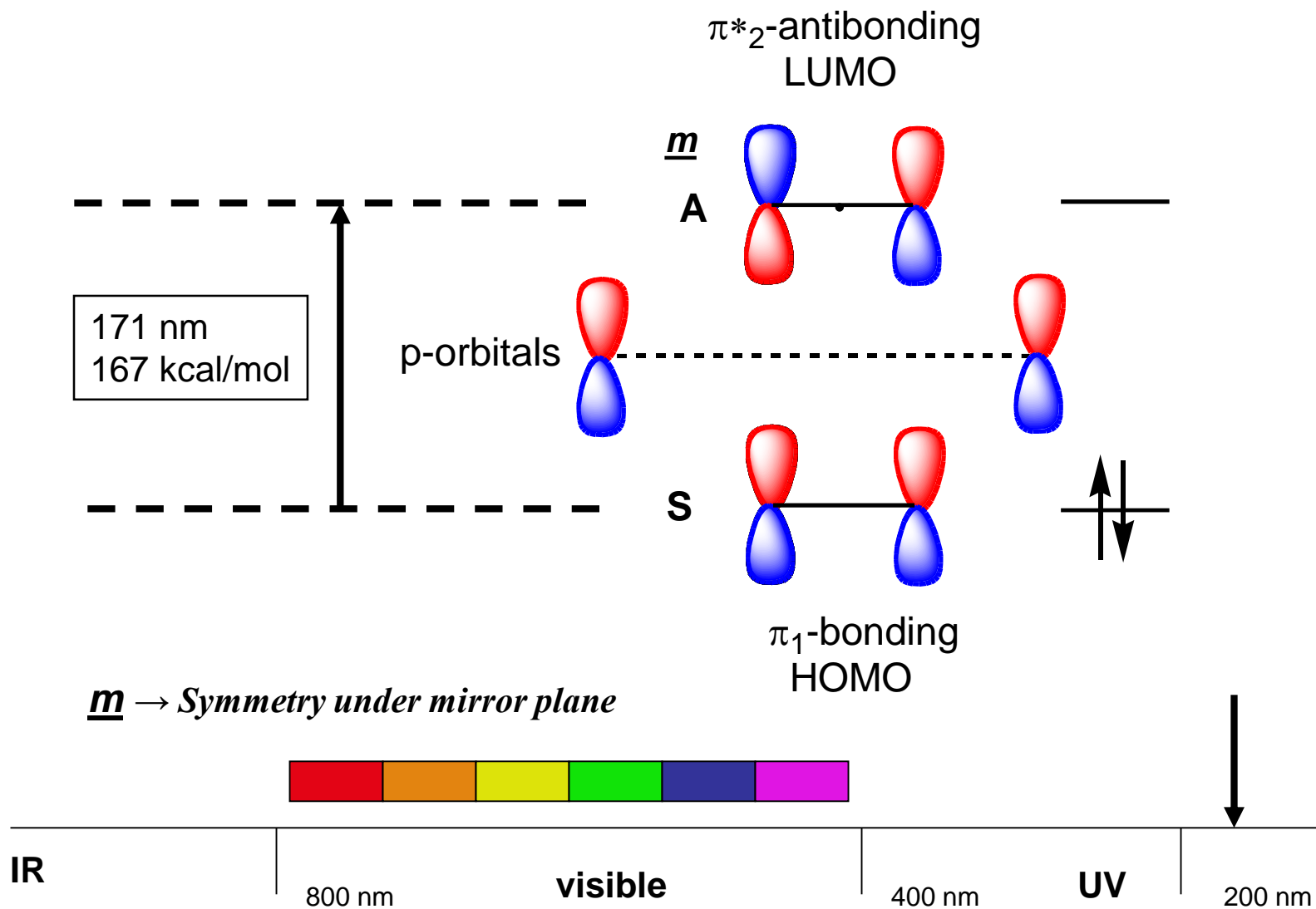
Electrocyclic reactions: Conrotatory and disrotatory motion, $4n$ and $(4n+2)$ systems, Cycloaddition reaction: $[2+2]$ and $[4+2]$ cycloaddition reaction, Cycloaddition of ketones, Secondary effects in $[4+2]$ cycloaddition. Stereochemical effects on rate of cycloaddition reaction, Diels-Alder reaction, 1,3-dipolar cycloaddition, Chelotropic reaction, The Nazarov reaction.

Sigmatropic rearrangement: Suprafacial and antarafacial shift involving H and carbon-moieties, Peripatetic cyclopropane bridge, Retention and inversion of configuration, $[3,3]$ -, $[1,5]$ -, $[2,3]$ -, $[4,5]$ -, $[5,5]$ -, and $[9,9]$ -Sigmatropic rearrangements, Claisen rearrangements (including Aza-Claisen, Ireland-Claisen), Cope rearrangements (including Oxy-Cope, Aza-Cope), Sommelet-Hauser rearrangements, Group transfer reaction, Ene reaction, Mislow - Evans rearrangement, Walk rearrangement.

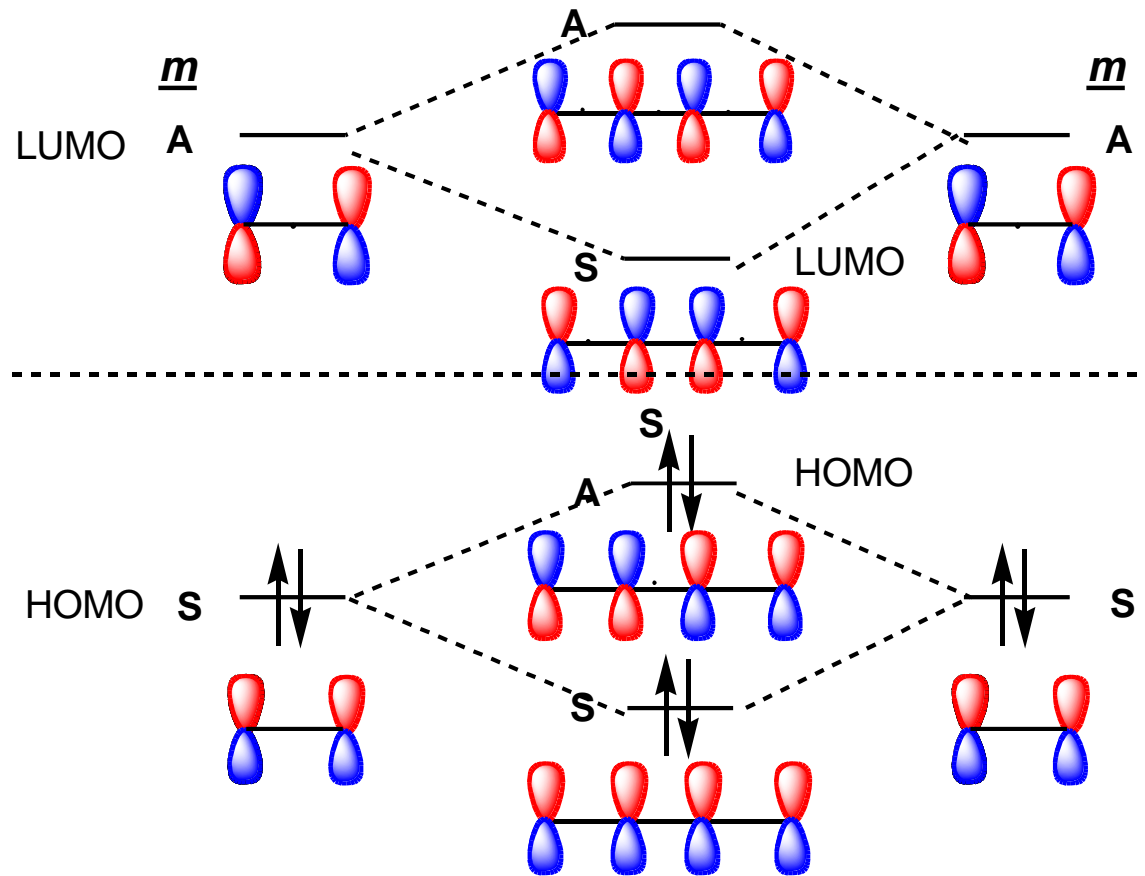
Coverage:

1. Molecular Orbital Symmetry
2. Frontier Orbitals of Ethylene, 1,3-Butadiene, 1, 3, 5-Hexatriene and Allyl System

π -Molecular Orbitals of Ethylene

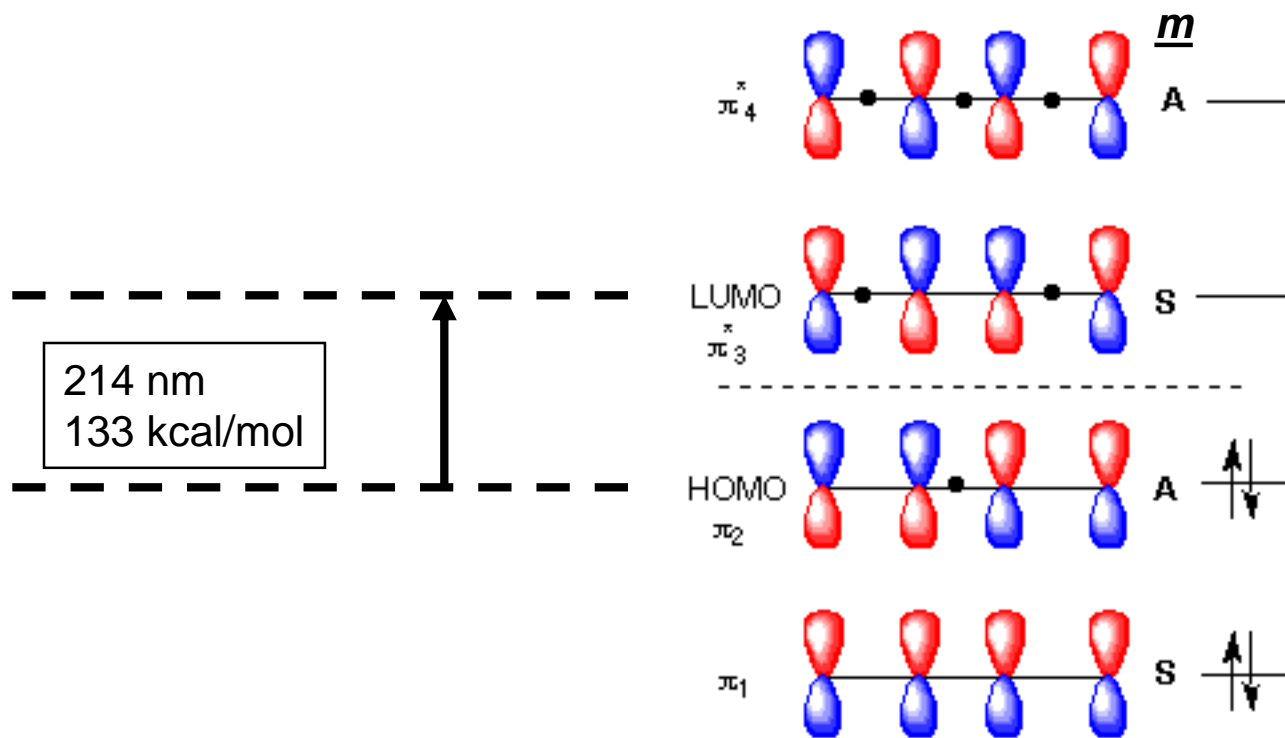


π -Molecular Orbitals of 1,3-Butadiene from Ethylene



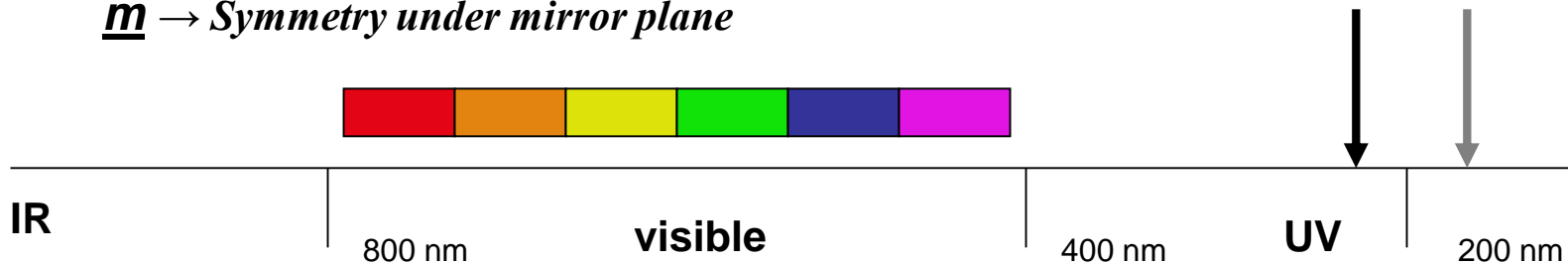
\underline{m} \rightarrow Symmetry under mirror plane

π -Molecular Orbitals of 1,3-Butadiene

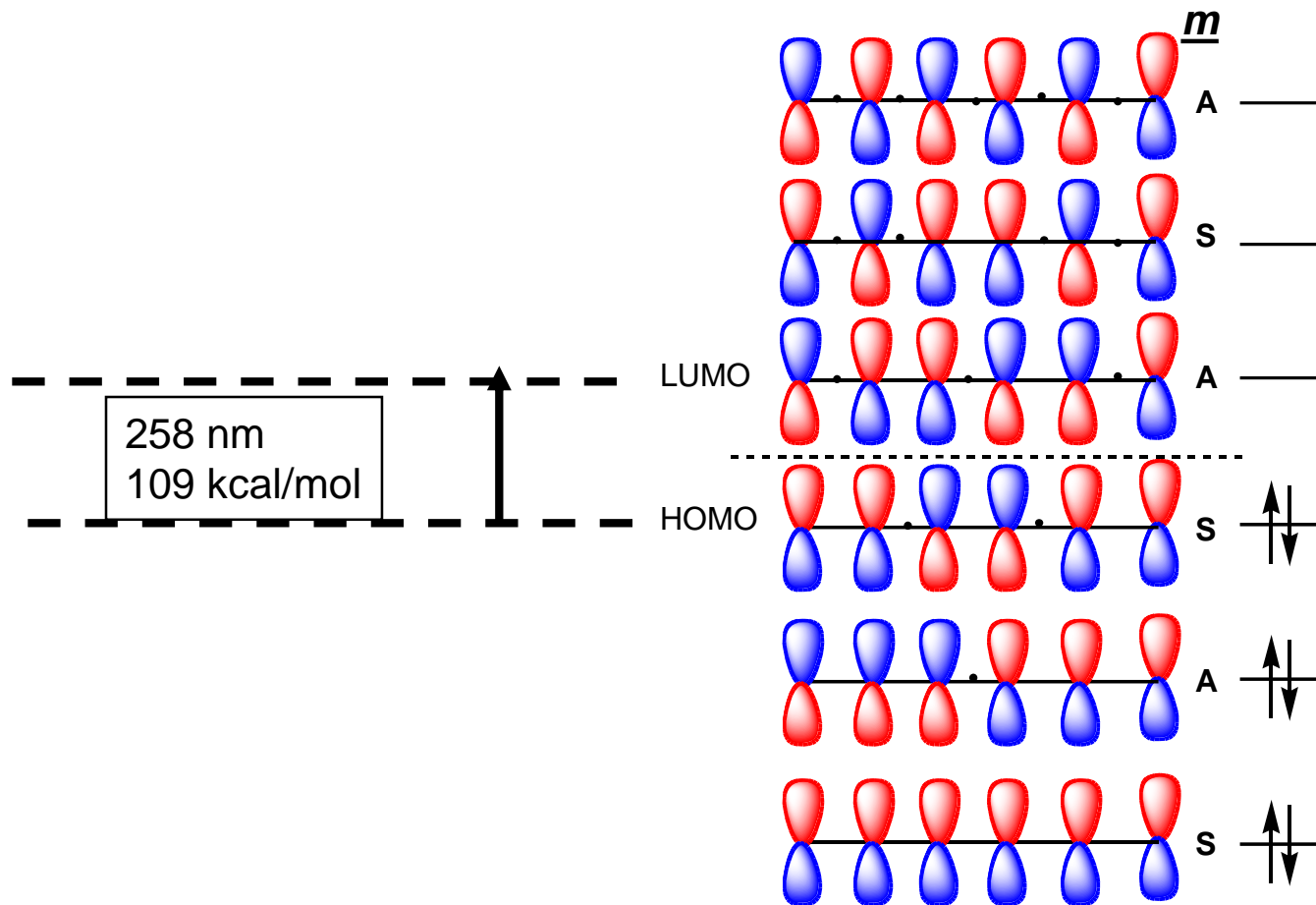


214 nm
133 kcal/mol

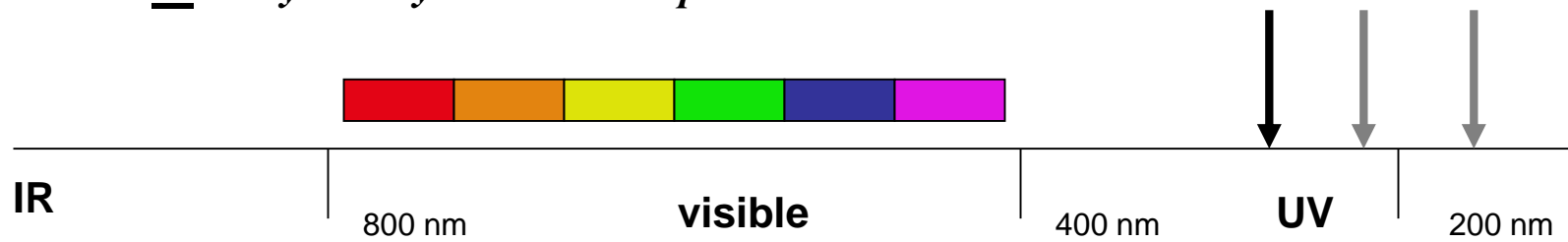
m \rightarrow Symmetry under mirror plane



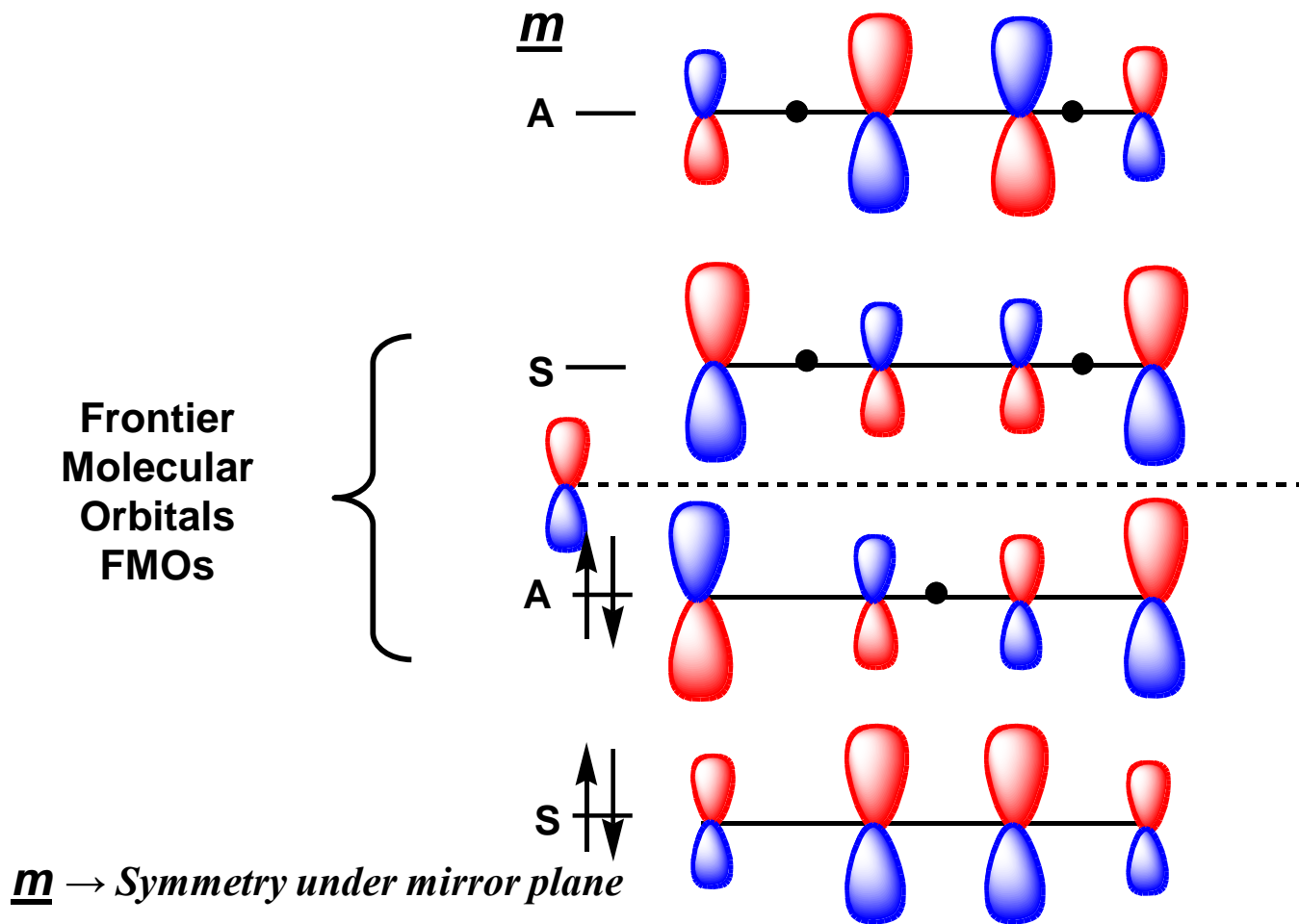
π -Molecular Orbitals of 1,3,5-Hexatriene



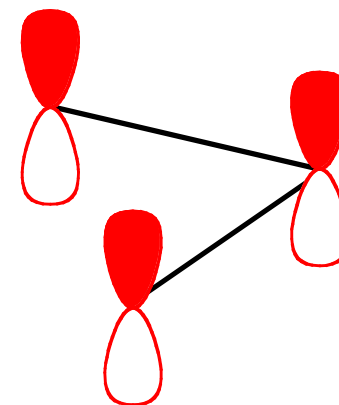
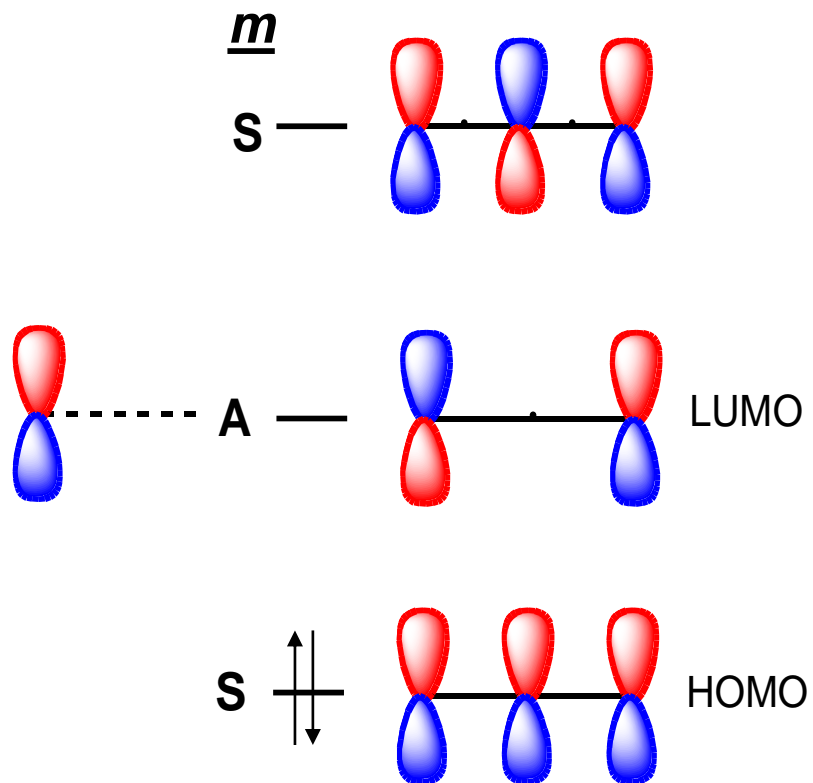
\underline{m} \rightarrow Symmetry under mirror plane



Butadiene: Orbital Coefficients

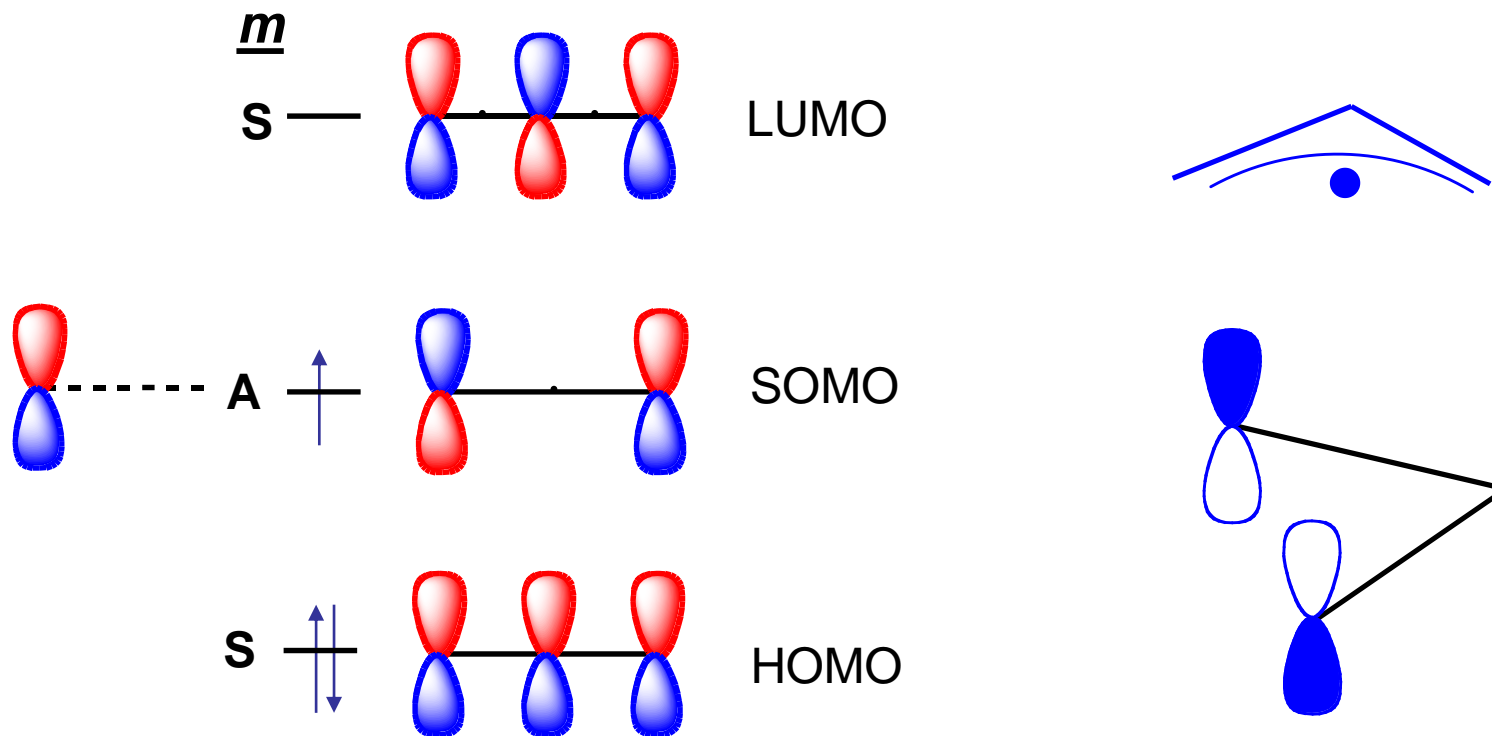


The Allylic System: Allyl Cation



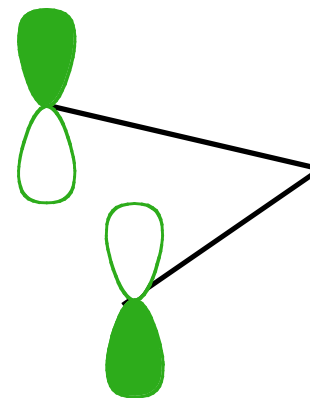
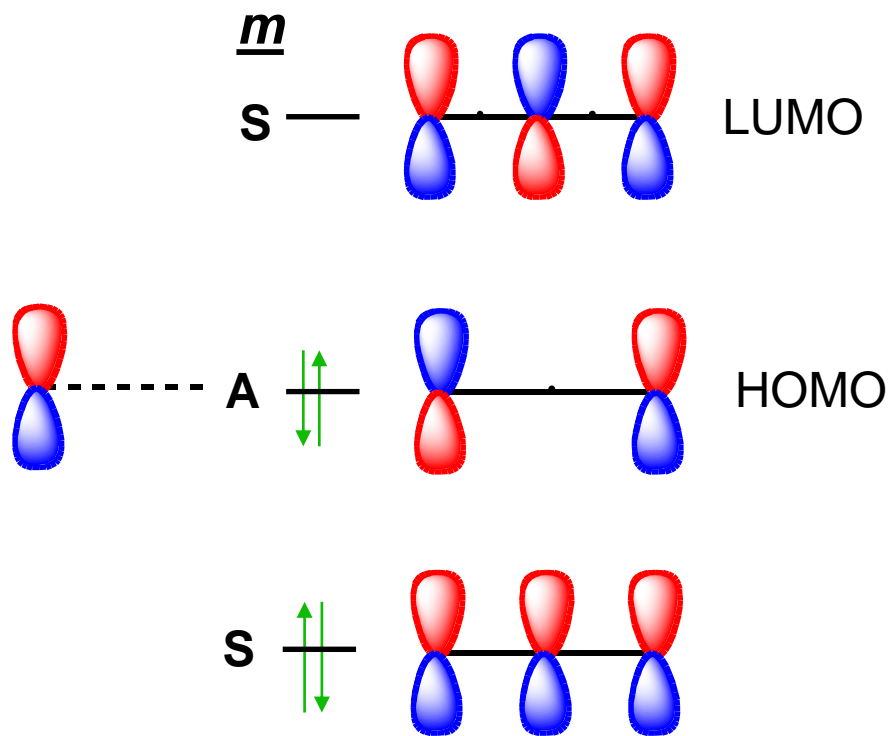
\underline{m} → Symmetry under mirror plane

The Allylic System: Allyl Radical



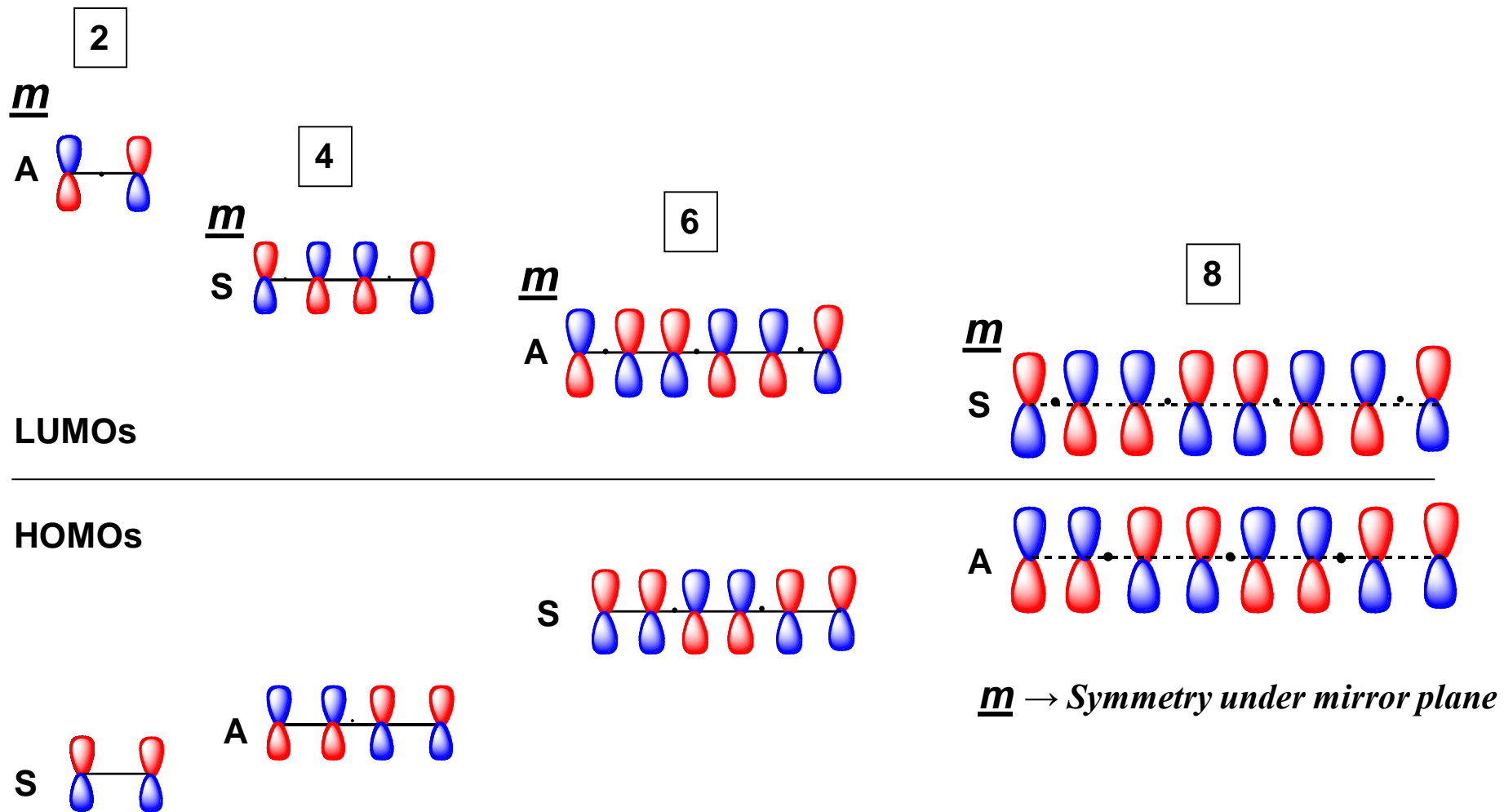
\underline{m} → Symmetry under mirror plane

The Allylic System: Allyl Anion

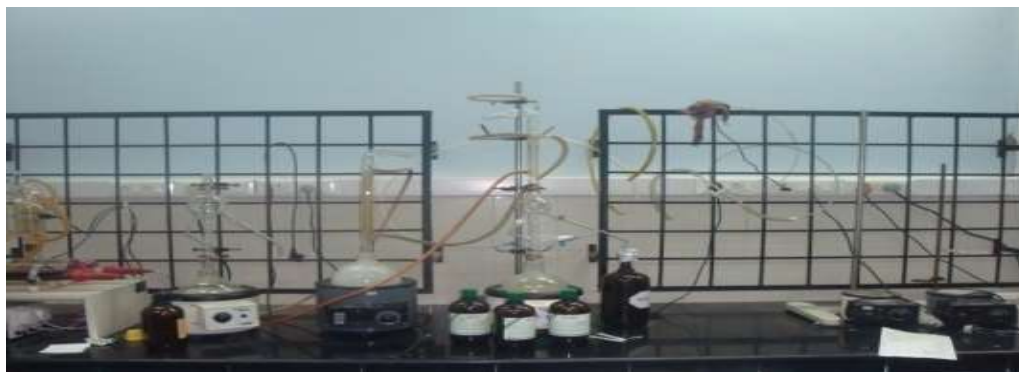


\underline{m} → Symmetry under mirror plane

Energy Gap Between HOMO and LUMOs



Thank You



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