

**B.Sc. Semester-VI
GroupA / DSE-4
Organic Synthesis**



**II. Pericyclic Reactions
10. Symmetry Control in Pericyclic Reactions**



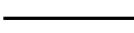
**Dr. Rajeev Ranjan
University Department of Chemistry
Dr. Shyama Prasad Mukherjee University,
Ranchi**

Pericyclic Reactions

• Polar react. (nucleophiles and electrophiles) $\text{Nu}^- : \quad \text{E}^{\oplus}$

• Radical react. $\text{R} \cdot \quad \cdot \text{R}'$

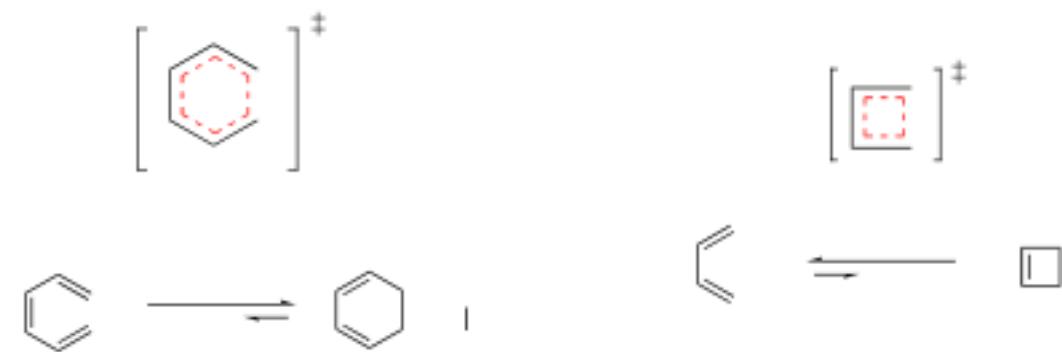
• Pericyclic react. (concerted, cyclic $\text{TS}^\#$)

• Electrocyclic react.  • Rearrangement of polyene

• Cycloadditions (i.e. Diels Alder)

• Termal (react. in ground state)
or photochemical
(react of exited state)

• Sigmatropic rearrangement

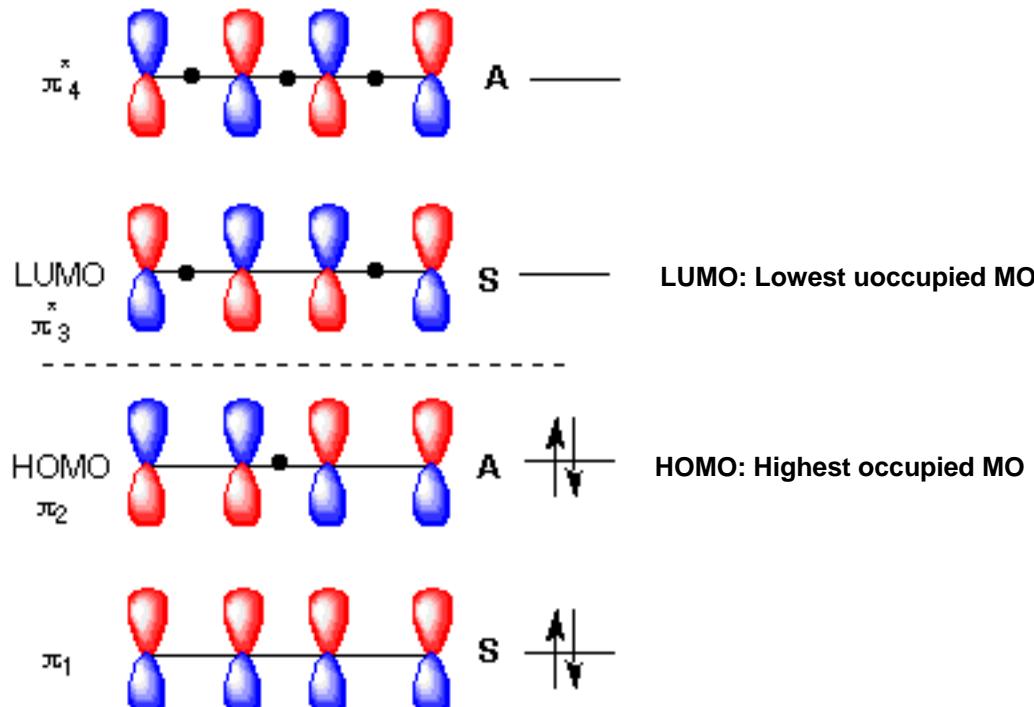


Symmetry Allowed Reaction

Woodward Hoffmann rules

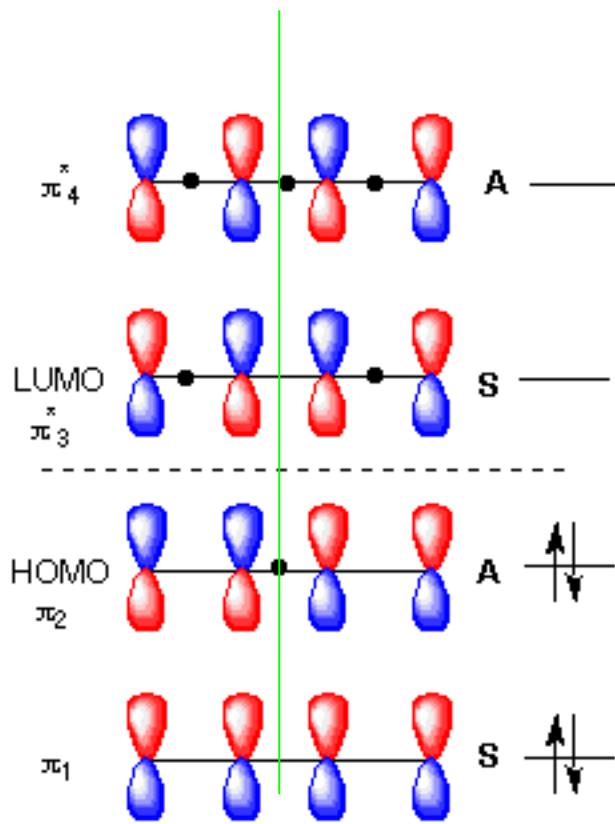
Symmetry in reactants are preserved during pericyclic react.

Results can generally be predicted just by looking at Front Orbitals (FMO; HOMO and LUMO) - Fukui



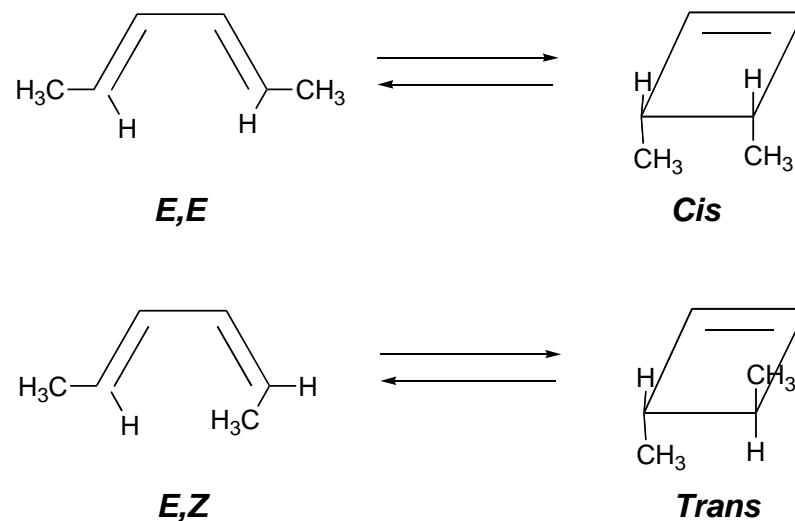
Molecular orbitals

1,3-butadiene

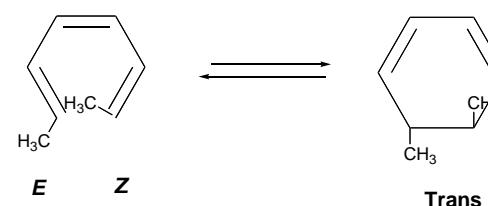
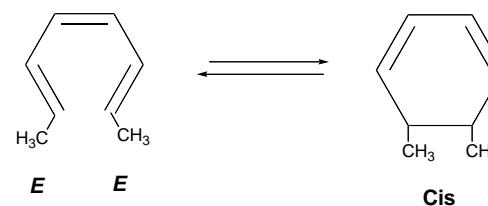
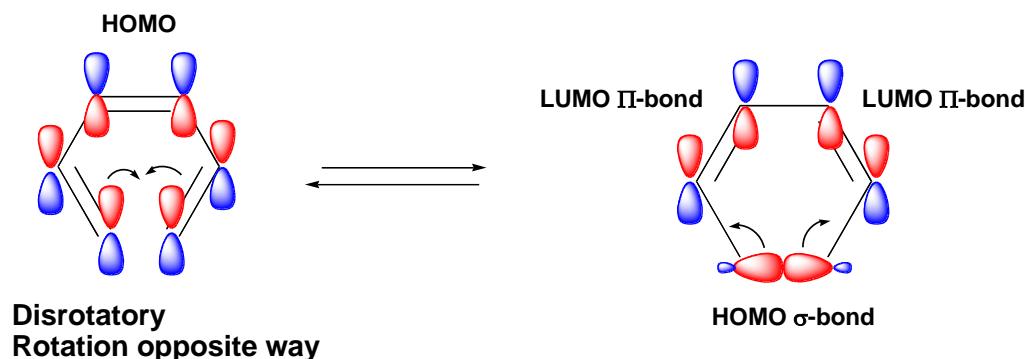
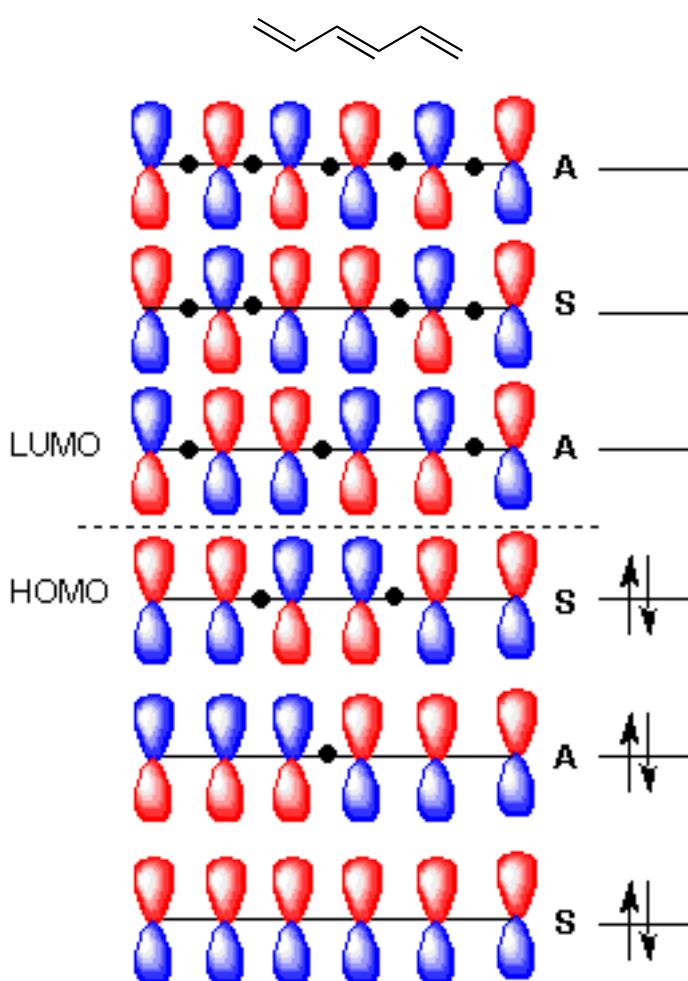


S: Symmetric
A: Antisym.

Stereospecific react.



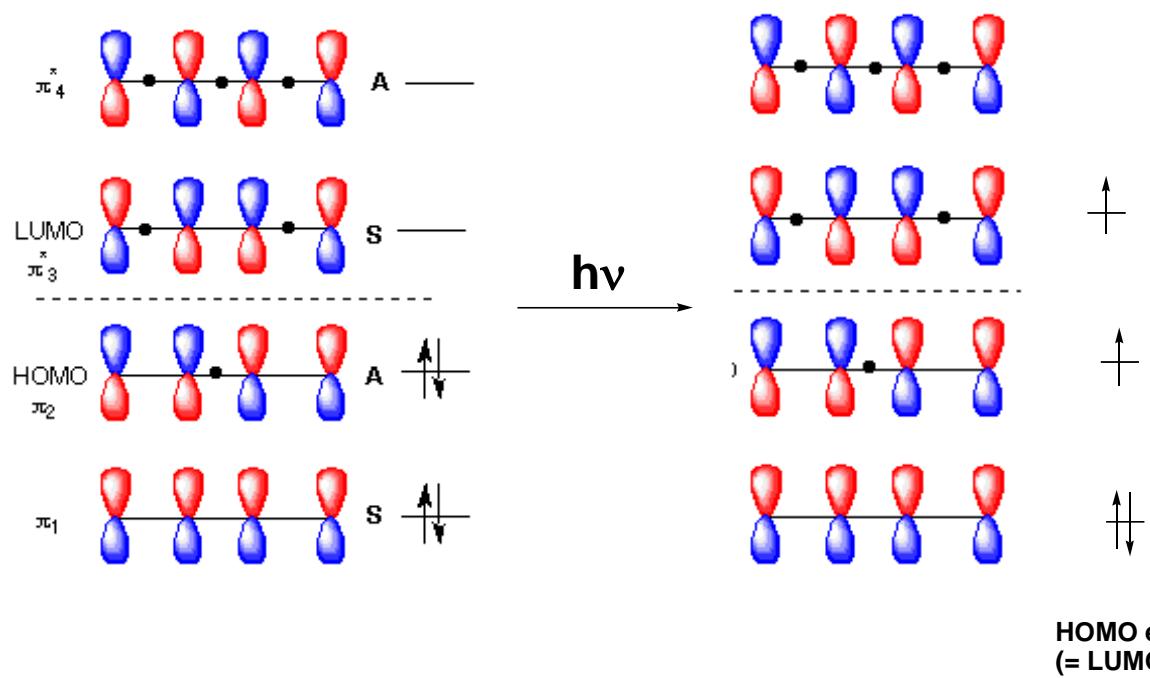
Molecular orbitals hexatriene



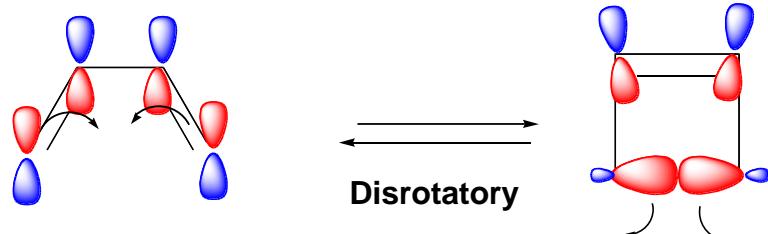
Symmetry allowed react

No. of electrons	Reactions in the ground state
$4n$	Conrotatory
$4n + 2$	Disrotatory

Photochemical electrocyclic reaction



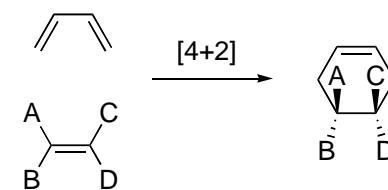
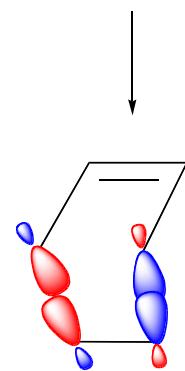
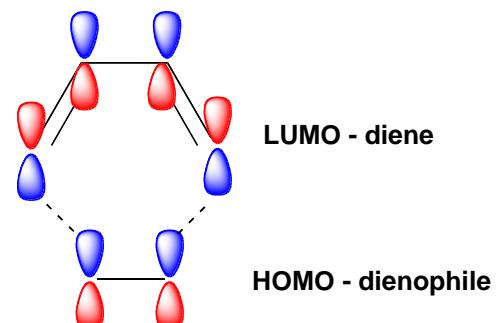
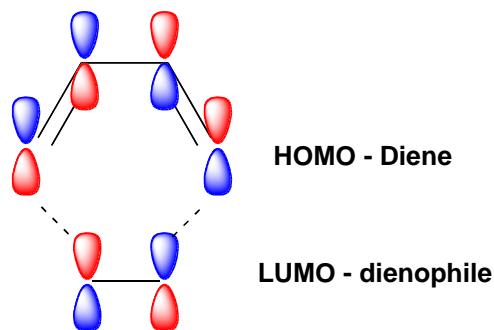
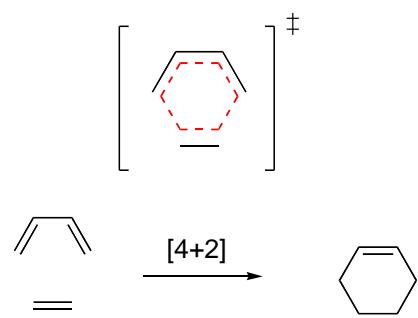
HOMO exitet state
(= LUMO ground state)



No. of electrons	Reactions in the ground state (termal)	Reactions in exited state (Photochem.)
4n	Conrotatory	Disrotatory
4n + 2	Disrotatory	Conrotatory

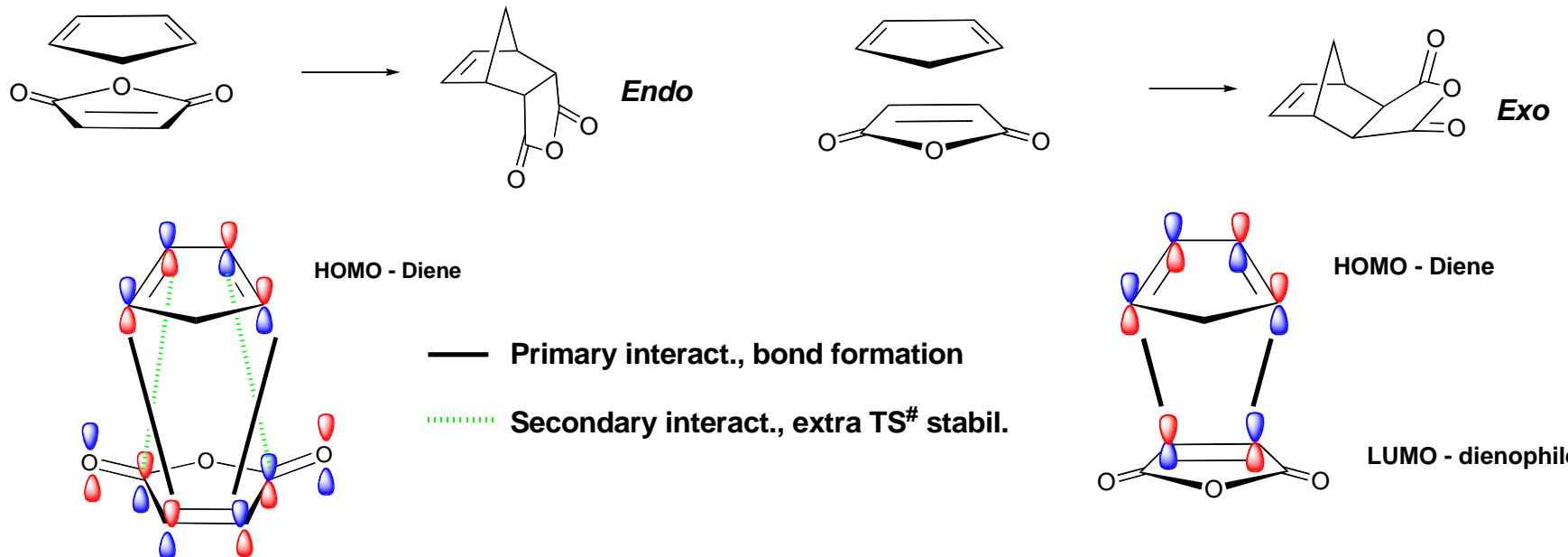
Cycloadditions (i.e. Diels Alder Reaction)

Suprafacial cycloadd.



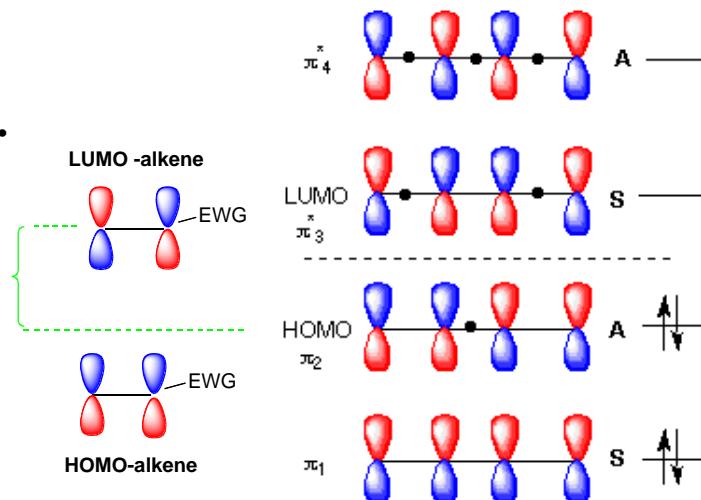
Stereospecific react.

endo - *exo* selectivity

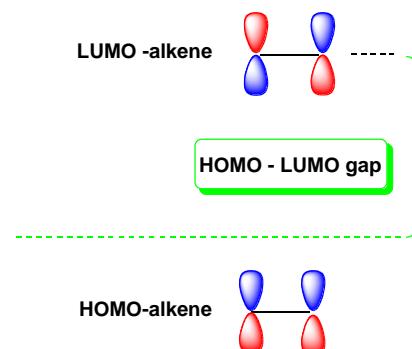


Normal electron demand DA - Electron poor dienophile (Michael accept.)

Michael accept.
Lower LUMO

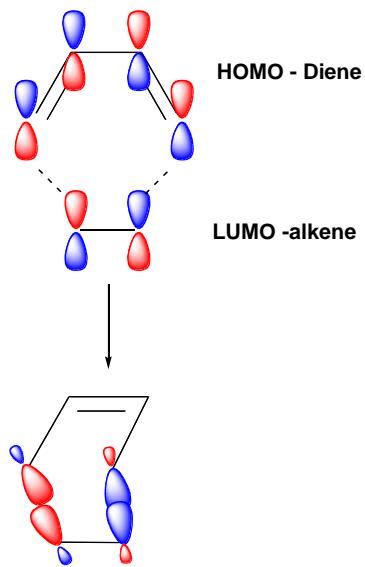


Ethene etc, very low react.

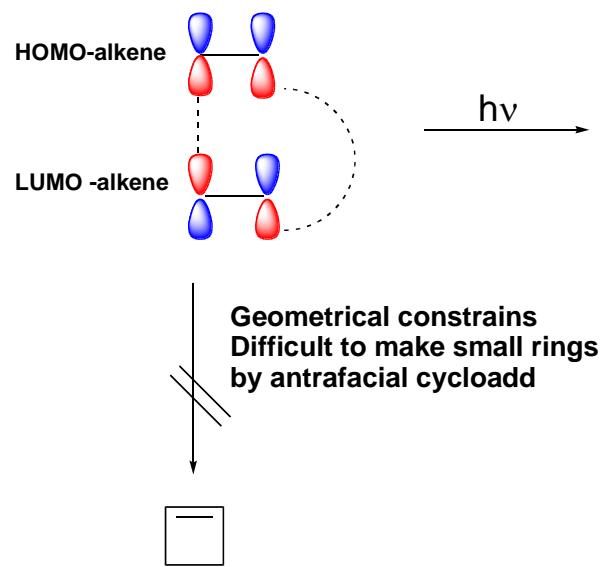


[2+2] Cycloadditions

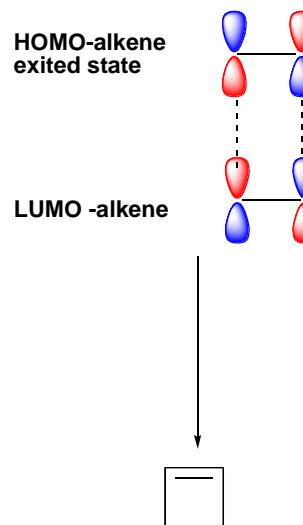
Suprafacial cycloadd.



Antarafacial cycloadd. (termal cond.)



Suprafacial cycloadd. (photochem. cond.)



No. of electrons
4n

Reactions in the ground state
(termal)

$4n + 2$

Antarafacial

Suprafacial

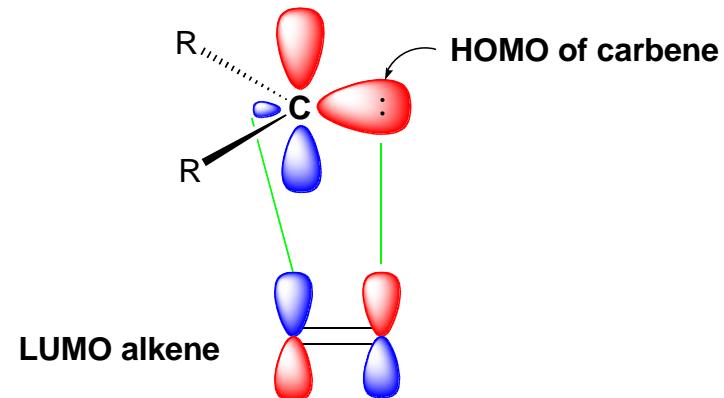
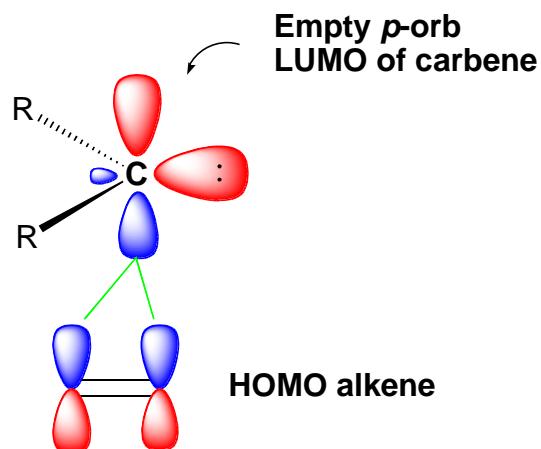
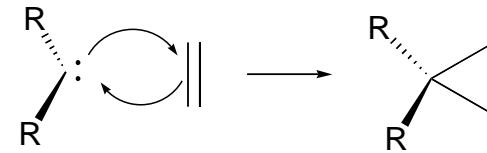
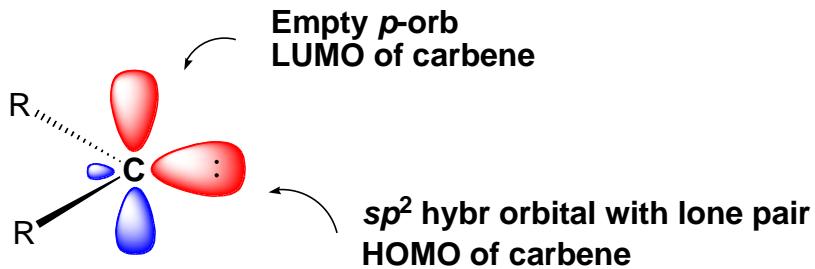
Reactions in exited state
(Photochem.)

Suprafacial

Antarafacial

Carbene Cycloadditions

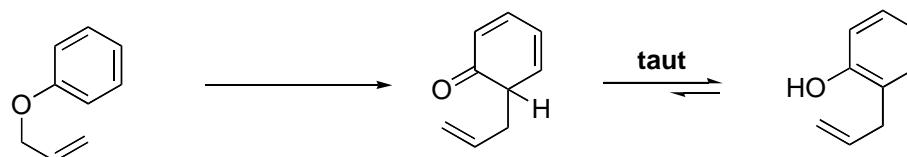
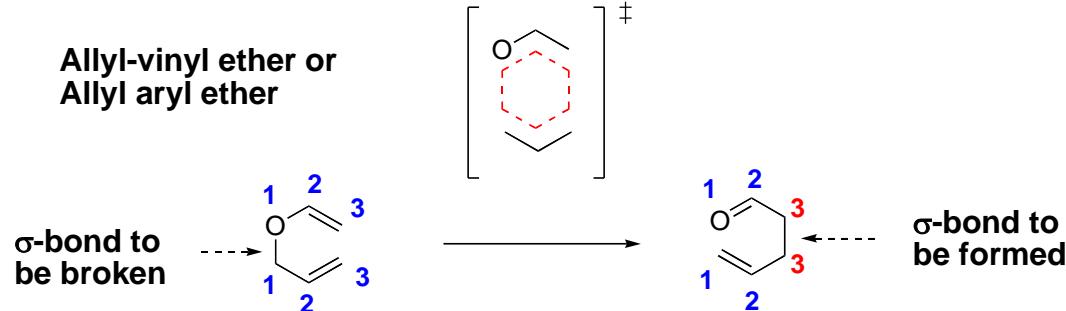
Singlet Carbene



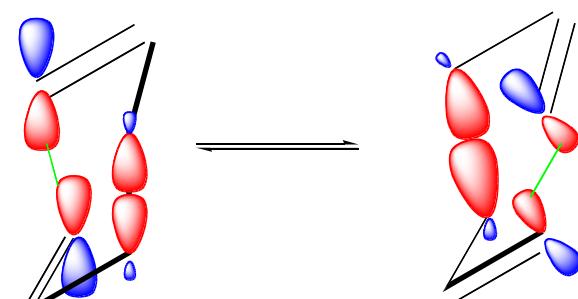
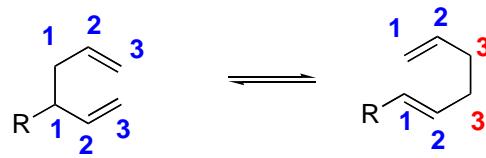
Sigmatropic Rearrangements

[3,3] Rearrangements; Claisen rearrang. etc.

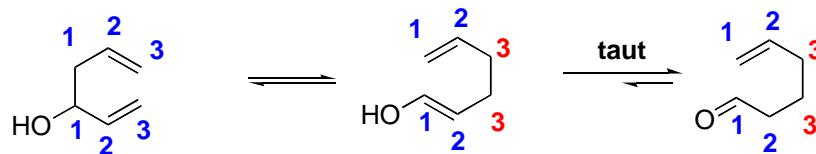
Claisen rearrangement



Cope rearrangement

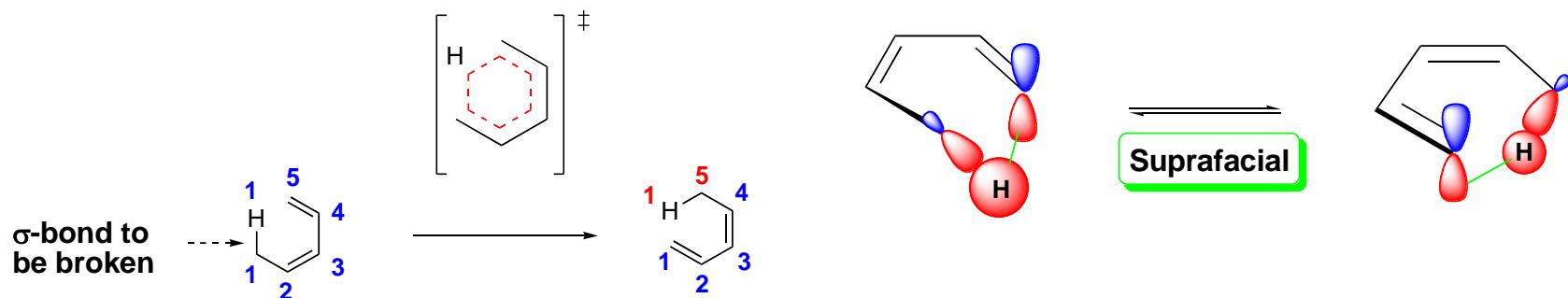


Oxy-Cope rearrangement



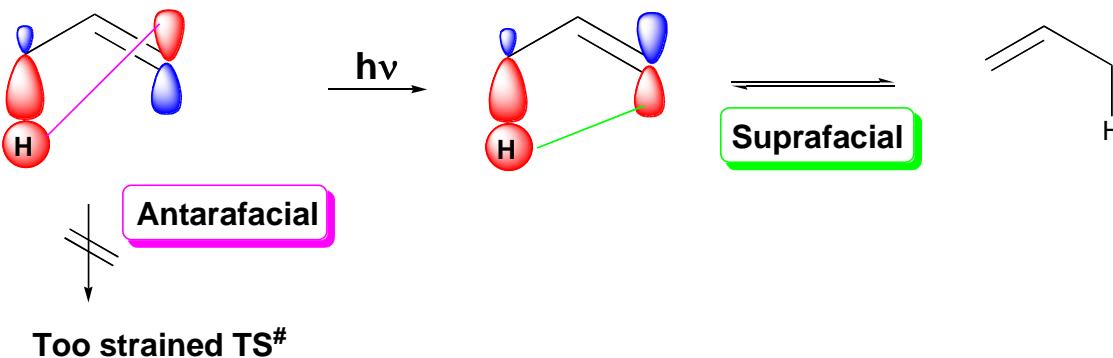
Suprafacial

[1,5] Rearrangement (H-shift)



No. of electrons	Reactions in the ground state (termal)	Reactions in excited state (Photochem.)
$4n$	Antarafacial	Suprafacial
$4n + 2$	Suprafacial	Antarafacial

[1,3] Rearrangement (H-shift)



Thank You



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